

**Application Note** 

# Gas Analysis System using Full Vacuum Type FT/IR (Trace amount of H<sub>2</sub>O in N<sub>2</sub> gas)

## Introduction

The analysis of low concentration gases using long pathlength gas cells has been reported previously, however, it can be difficult to accurately quantitate gases at low concentrations when the absorption peaks of the target gas overlap with the peaks of atmospheric water vapor or, when the target gas itself is H<sub>2</sub>O or CO<sub>2</sub>, which are presnet in the atmosphere. This is because even with a vacuum FT-IR instrument, it was not possible to evacuate the sample chamber with the long pathlength cell present. The full vacuum gas analysis system introduced in this application note has a special gas cell integrated into the sample chamber, allowing the entire light path to remain under full vacuum with the gas cell in place. This full vacuum model makes it possible to quantitate concentrations of H<sub>2</sub>O even at the 0.1 ppm level.

### **Instrument and Measurement**

For the measurement, there are separate vacuum lines for evacuation of the FT-IR instrument and the gas cell. with independent evacuation capability for the separate components.

low concentrations of the gas samples.

## **Cell specification**

Cell type: Pathlength : Cell body: Mirror material: Mirror surface: O-ring: Window: Heating: Cell capacity: Gas in/output port: Multi-pass 'White' cell 10 meters Stainless steel Stainless steel Gold coated Viton CaF<sub>2</sub> Possible, max. 100°C Approx. 2 L 1/4 inch VCR



Fig. 1 Full vacuum type FT-IR gas analysis system (10 meter cell)

#### Measurement example

The quantitation of a trace amount of water vapor (H<sub>2</sub>O) within nitrogen gas was attempted. Water vapor with a concentration of 100 ppm was diluted by monitoring the pressure gauge during dilution of the standard samples with concentrations of 0.0, 0.51, 1.02 and 2.48 ppm. Figure 2 shows the IR spectra of water vapor for each concentration. For the 0 ppm concentration, there was no absorption due to water vapor observed in the spectrum. The calibration curve in Figure 3 was created using the absorption peak at 1653 cm<sup>-1</sup>.

> JASCO, USA 8649 Commerce Drive Easton, Maryland 21601





Fig. 2 Expanded region for spectra of low concentration H<sub>2</sub>O



Fig. 3 H<sub>2</sub>O calibration curve